

Enhanced photocatalytic Performance of TiO₂ nano-rods and nano-crystals using graphene as a cocatalyst

Long Zhang^{1,2*}, Yanfeng Ma³, Yongsheng Chen³

¹Institute of Chemical Materials, China Academy of Engineering Physics, No. 64, Mianshan Road, Mianyang City, China.

²Sichuan New Material Research Center, No. 20, yuan yi street, Mianyang, China.

³Key Laboratory of Functional Polymer Materials and Center for Nanoscale Science & Technology, Institute of Polymer Chemistry, College of Chemistry, Nankai University, Tianjin, China
zhanglong@caep.cn

Abstract

Different graphene/TiO₂ nanocomposites including graphene/TiO₂ nanorods and graphene/TiO₂ nanocrystals were prepared by using simple solution methods and further annealing process. The effective anchoring of TiO₂ nanorods or nanocrystals on the whole graphene sheets were demonstrated by transmission electron microscopy, UV-Vis and X-Ray Diffraction. It is demonstrated that not only compared with the commercial TiO₂ (P25), both pure TiO₂ nanorods and TiO₂ nanocrystals exhibit higher photocatalytic activities on the water splitting reaction under UV light irradiation, but also the introduction of small amount graphene in each kind of TiO₂ results in greatly improved photocatalytic activities for their graphene/TiO₂ nanocomposites. The large enhancement of photocatalytic properties can be attributed to the shortened electron transferring distance from size-reduced TiO₂ and the effective charges-separation assisted by graphene.

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Figures

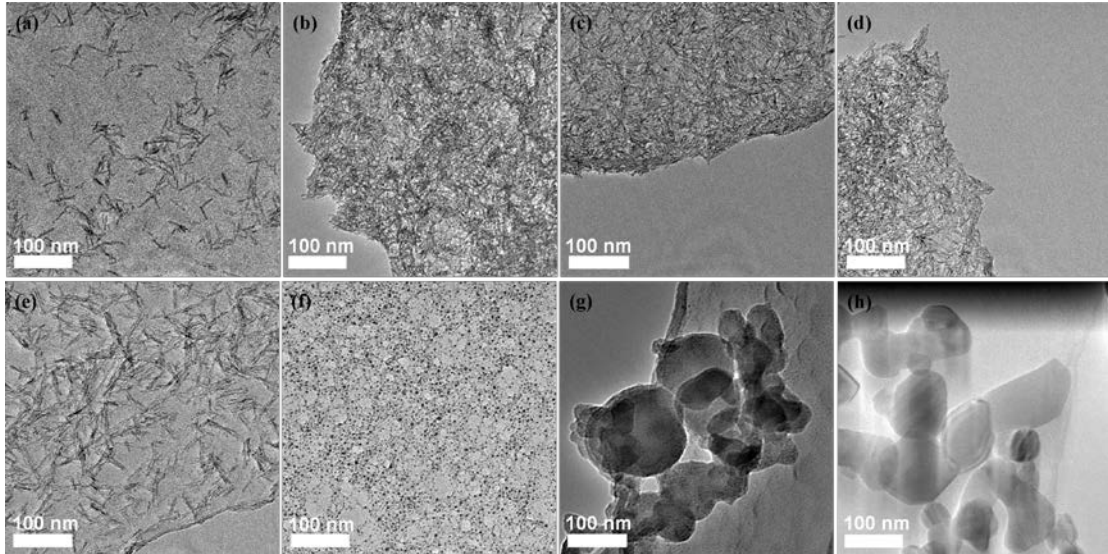


Figure 1 Typical TEM images of (a) pure TiO₂ nanorods and (b-e) graphene/TiO₂ nanorod nanocomposites with 1 wt%, 5 wt%, 10 wt% and 20 wt% of graphene, respectively. (f) graphene/TiO₂ nanocrystals nanocomposites with 10 wt% graphene. (g) Pure TiO₂ nanocrystals. (h) P25 (commercial TiO₂). The introduction of graphene during the preparation process of TiO₂ nanorods or nanocrystals has played an important role in preventing the aggregation of TiO₂ nanorods or nanocrystals, leading to much smaller diameters of TiO₂ nanorods or nanocrystals.

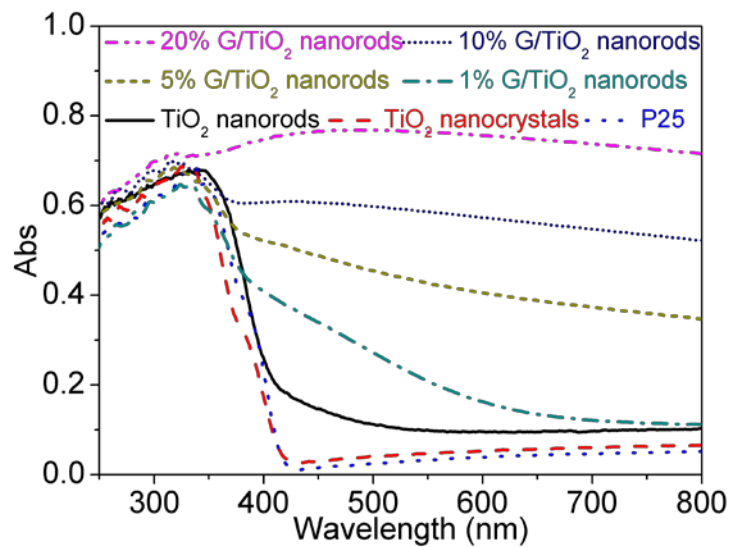


Figure 2 UV-visible-NIR absorption spectra of graphene/TiO₂ nanocrystals and nanorods. P25 was also tested for comparison. All the samples exhibit a characteristic peak for the TiO₂.

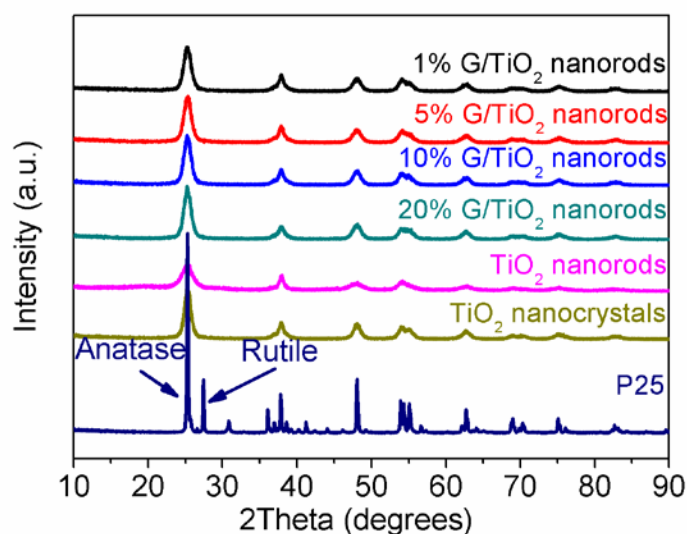


Figure 3 XRD analysis of graphene/TiO₂ nanocrystals and nanorods. The prepared TiO₂ nanorods and nanocrystals in the graphene/TiO₂ exhibit anatase TiO₂ phase, while P25 shows the diffraction peaks attributed to both anatase and rutile TiO₂ phases.

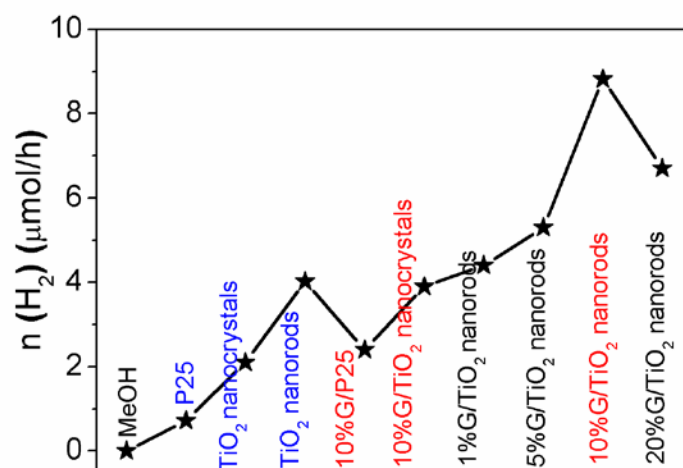


Figure 4 Photocatalytic performance which demonstrated the amount of hydrogen evolution under 1 h irradiation of UV-vis light for pure P25, TiO₂ nanorods, TiO₂ nanocrystals and different graphene/TiO₂ nanocomposites. Not only both pure TiO₂ nanorods and TiO₂ nanocrystals exhibit higher photocatalytic activities than pure P25, but also the introduction of small amount graphene in TiO₂ nanorods or TiO₂ nanocrystals results in greatly improved photocatalytic activities for their graphene/TiO₂ nanocomposites.